

Answers For Chapter One Chemical Foundations Exercises

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Answers For Chapter One Chemical

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Answer. (a) Corrosion ljt is a chemical process of decay of metal when they are exposed to moisture. (b) Rusting Corrosion of iron is called rusting. It is a slow oxidation reaction which takes place in the presence of moisture. We hope the NCERT Solutions for Class 10 Science Chapter 1 Chemical Reactions and Equations help you.

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Ans: The first chapter of Class 10 Science Chemical Reactions and Equations is an interesting chapter. It is also one of the most conceptual chapters of the science textbook which requires the right knowledge and understanding to get the answers right in exams.

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NCERT Solutions Class 11 Chemistry Chemistry Lab Manual Chemistry Sample Papers. NCERT TEXTBOOK QUESTIONS SOLVED. Question 1. Calculate the molecular mass of the following: (i) H₂O (ii) CO₂ (iii) CH₄
Answer: (i) Molecular mass of H₂O = 2(1.008 amu) + 16.00 amu = 18.016 amu (ii) Molecular mass of CO₂ = 12.01 amu + 2 x 16.00 amu = 44.01 amu (iii) Molecular mass of CH₄ = 12.01 amu + 4 (1...

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Here we have given HOTS Questions for Class 10 Science Chapter 1 Chemical Reactions and Equations. Question 1. Study the given diagram and answer the following questions : (a) Write the chemical reaction involved in the process. (b) Mention the colour of : copper powder and; the substance formed after heating it. (c) How can we reverse the ...

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Express your answer to three significant figures and include the appropriate units. 3.54×10^4 g A block of metal has a width of 3.2 cm, a length of 17.1 cm, and height of 4.5 cm .

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Important Questions for CBSE Class 10 Science Chapter 1 ...

Hits: 13 EXERCISES:-(Page:14-16) 1.Which of the reactions about the reaction below are incorrect? $2\text{PbO}(s)+\text{C}(s)\rightarrow 2\text{Pb}(s)+\text{CO}_2(g)$ a) Lead is getting oxidised b) Carbon dioxide is getting oxidised c) Carbon is getting oxidised d) Lead oxide is getting reduced ANSWER- (a) 2. $\text{Fe}_2\text{O}_3+2\text{Al}\rightarrow \text{Al}_2\text{O}_3+2\text{Fe}$ The above reaction is an example of a a) Combination reaction b) Double displacement reaction [...]

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