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aerospace engineers. About the Author ROBERT D. ZUCKER, P H D, is Professor Emeritus of Aeronautics and Astronautics at the Naval Postgraduate School in Monterey, California.

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The form of the perfect gas equation normally used in gas dynamics is $p = RT$ (1.13) where p absolute pressure lbf/ft²

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or N/m^2 density lbm/ft^3 or kg/m^3 T absolute temperature R or K R individual gas constant ft-lbf/lbm-R or N m/kg K The individual gas constant is found in the English Engineering system by dividing 1545 by the molecular mass of the gas chemical constituents.

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You don't need much background to enter the fascinating world of gas dynamics. However, it will be assumed that you have been exposed to college-level courses in calculus and thermodynamics. Specifically, you are expected to know: 1. Simple differentiation and integration 2. The meaning of a partial derivative 3. The significance of a dot product

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Fundamentals of gas dynamics by Robert D. Zucker, 2002, John Wiley & Sons, Ltd. edition, Electronic resource in English

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Fundamentals of Gas Dynamics, 2nd Edition. Robert D. Zucker, Oscar Biblarz. ISBN: 978-0-471-05967-7 August 2002 512 Pages. E-Book. Starting at just \$133.99. Print. Starting at just \$166.50. E-Book. \$133.99. Hardcover. \$166.50. Read an Excerpt Excerpt 1: (PDF) Download Product Flyer

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